

Year 12	Autumn 1 & 2	Between Spring 1 & Summer 2	
Торіс	Foundations in Chemistry	Periodic Table & Energy	
Key Concepts	 Module 2 Chapters 2-6: Atomic structure and isotopes Relative mass Formulae and equations Amount of substance and the mole Determination of formulae Moles and volumes Reacting quantities Acids, bases and neutralisation Acid-base titrations Redox Electron structure Ionic bonding and structure Covalent bonding Shapes of molecules and ions Electronegativity and polarity Intermolecular forces Hydrogen Bonding 	 Module 3 Chapters 7-10: The Periodic Table Ionisation energies Periodic trends in bonding and structure Group 2 The halogens Qualitative analysis Enthalpy changes Measuring enthalpy changes Bond enthalpies Hess' law and enthalpy cycles Reaction rates Catalysts The Boltzmann distribution Dynamic equilibrium and le Chatelier's principle The equilibrium constant Kc 	Module 4 Chap Properties of Reactions of The chemistr Organohalog Practical tec Synthetic rou Mass spectro Infrared spec
	Core Organic Chemistry	Physical Chemistry & Transition Ele	
	 Module 4 Chapters 11-13: Organic chemistry Nomenclature of organic compounds Representing the formulae of organic compounds Isomerism Introduction to reaction mechanisms Properties of alkanes Chemical reactions of alkanes Properties of alkenes Stereoisomerism Reactions of alkenes Electrophilic addition of alkenes 	 Module 5 Chapters 18-19: Orders, rate equations & rate constants Concentration-time graphs Rate-concentration graphs Rate-determining step Rate constants and temperature The equilibrium constant Kc The equilibrium constant Kp Controlling the position of equilibrium 	 Module 5 Ch Bronsted-Low The pH scale The acid diss The pH of we pH and stron Buffer solutio Buffer solutio Neutralisation



Core Organic Chemistry

napters 14-17: es of alcohols ns of alcohols mistry of the haloalkanes nalogen compounds in the environment techniques in organic chemistry routes ctrometry spectroscopy

ements

5 Chapters 20-21: -Lowry acids and bases cale and strong acids dissociation constant Ka f weak acids trong bases lutions lutions in the body



Year	13

Year 13	Autumn & Spring Terms		
Торіс	Physical Chemistry & Transition Elements	Organic Chemistry & Analysis	
Key Concepts	 Module 5 Chapters 22-24: Lattice enthalpy Enthalpy changes in solution Factors affecting lattice enthalpy and hydration Entropy Free Energy Redox reactions Manganate (VII) redox titrations lodine/ thiosulphate redox titrations D-block elements Formation & shapes of complex ions Stereoisomerism in complex ions Ligand substitution & precipitation Redox & qualitative analysis 	 Module 6 Chapters 25-29 Introducing benzene Electrophilic substitution reactions of benzene The chemistry of phenol Distribution and directing groups Carbonyl compounds Identifying aldehydes and ketones Carboxylic acids Carboxylic derivatives Amines Amino acids, amides and chirality Condensation polymers Carbon-carbon bond formation Further practical techniques Further synthetic routes Chromatography and functional group analysis Nuclear magnetic resonance (NMR) spectroscopy Proton NMR spectra Interpreting NMR spectra Combined Techniques 	Revise



Summer Term

Revision & Exams

se for (and take) the A Level Chemistry exams





All students will sit an assessment and a mock examination in Year 12 and two mock examinations in Year 13.

	Year 12		Year 13		
	Assessment	Mock Exam	Mock Exam	Mock Exam	Revision Resources
	Autumn Term	Summer Term	Autumn Term	Spring Term	Kennet Resources
Style of Assessment	Multiple-choice and longer answer questions	Multiple-choice and longer answer questions	Multiple-choice and longer answer questions	Multiple-choice and longer answer questions	 Core Questions Knowledge Organisers
Topics Assessed	 All topics covered: Electrons & bonding Shapes of molecules & intermolecular forces Atoms, ions & compounds Amount of substance Acids and redox Basic concepts of organic chemistry Alkanes Alkenes 	 Electrons & bonding Shapes of molecules & intermolecular forces Atoms, ions & compounds Amount of substance Acids and redox Basic concepts of organic chemistry Periodicity Enthalpy Rates Alkanes Alkenes Alcohols Haloalkanes Spectroscopy (1) 	 Part 1: Summative of Year 12 topics Part 2: Rates of reaction Equilibria Acids, bases & pH Aromatic chemistry Carbonyl chemistry 	All topics covered over the course.	 Learning Habits External Resources www.physicsand mathstutor.com www.kerboodle.com You can also find additional revision material on Frog

